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Acquisition and Analysis of Data in a Pressurized Entrained-Flow Coal Gasifier for the Purposes of Simulation Validation

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Outline

- Introduction
- Background coal gasification research
- U. Utah pilot-scale coal gasifier
- Types of data available for validation
- Performance issues
- Uncertainty considerations
- Conclusions



Introduction

- Industrial-scale coal gasifiers are primarily pressurized, O2-blown, entrained-flow variety
- Cost of gasification systems provides strong incentive to optimize using computational simulation
- Access to gasifiers for acquisition of validation data is challenging



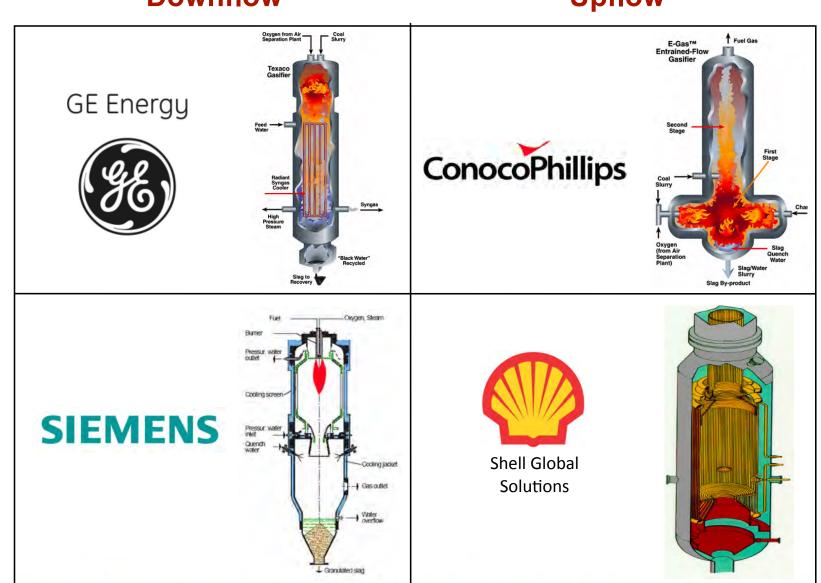
Pressurized O₂-Blown Entrained-Flow Gasifiers

Downflow

Upflow

Refractory-Lined





Challenges of Validation Data Acquisition

High temperature

- 1300-1500°C at reactor exit
- In excess of 2000°C within oxy-coal flame

High pressure

- IGCC application typically 25-30 atm (400 psi)
- Chemicals / fuel production 70+ atm (1000+ psi)

Corrosive environment

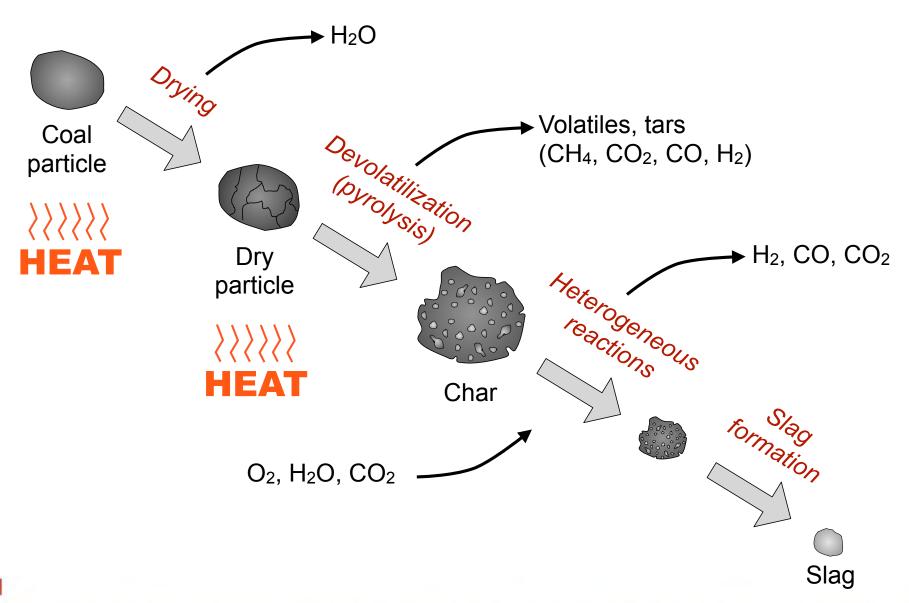
- Reducing environment
- Gaseous sulfur species (H₂S, COS)
- Molten coal slag

Consequences

- Crossing pressure boundary for gas sampling creates safety concerns
- Thermocouples typically last only a few days



Fundamental Coal Gasification Studies





Experimental Evaluation of Coal Conversion

- Drop tube (entrained-flow) furnaces
 - Pyrolysis yields
 - Char gasification kinetics
 - Physical transformations of coal particles
- Wire mesh heaters
 - Pyrolysis yields
- Thermogravimetric analyzers (TGAs)
 - Heterogeneous char gasification kinetics
- Mini-gasifiers
 - Electrically heated
 - Gases (CO₂, O₂) supplied from laboratory cylinders





"Small" versus "Big"

- Fundamental Studies ("small")
 - Up to perhaps 2 kg/day in entrained-flow reactors
 - Bottled gases
 - Electrically heated
- Commercial-Scale Systems ("big")
 - Hundreds of tons of coal (petcoke) per day
 - Oxygen-blown, with all associated mess
 - Difficult to access
- Need "medium" scale system to bridge this gap of 5 orders of magnitude



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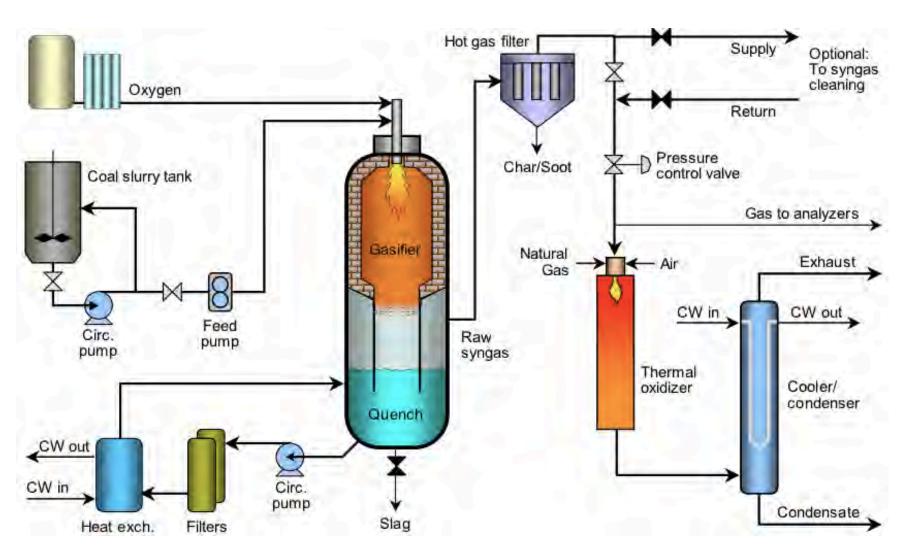


Bridging the Gap: UofU Gasifier

- Designed to operate like a "large" system
 - No electrical heating
 - Only inputs are oxygen and coal (slurry)
 - Similar in design to a GE gasifier
- Accessible like a "small" system
 - Reactor "stretched out" to decrease diameter and allow sampling at multiple residence times
 - Several (six) sampling ports down length of reactor
 - Six thermocouples for temperature measurement

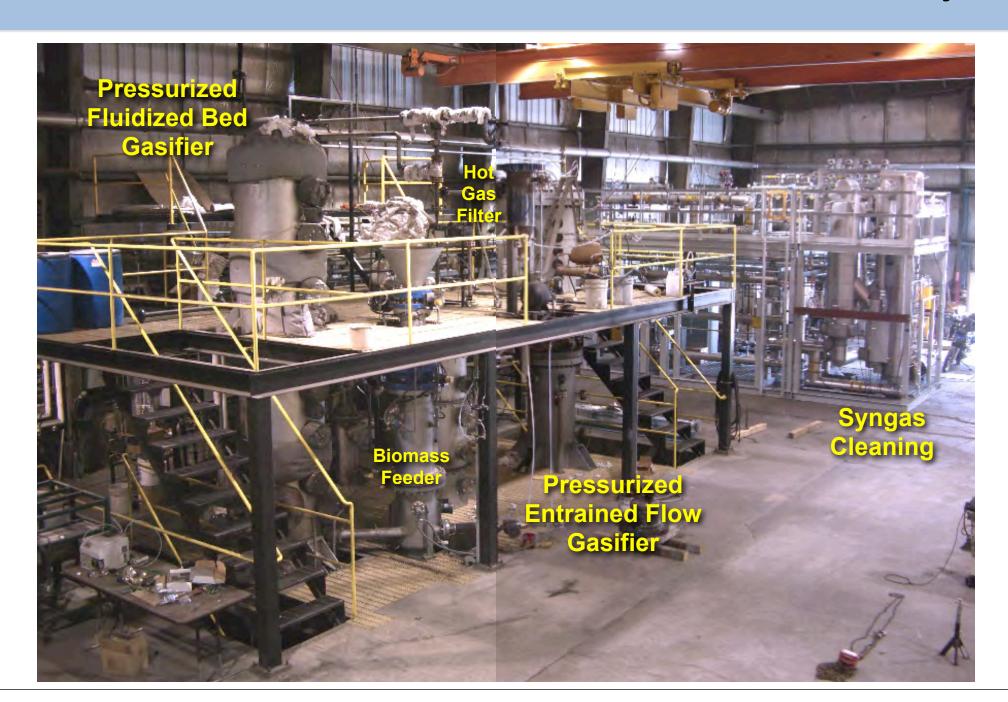


Gasifier System Schematic





Gasification Research Laboratory



Entrained-Flow Gasifier



Oxygen Supply System

- On-site oxygen tank
 - 6,000 gallons / 20 tons
 - Serves gasification and oxy-fuel systems
- "Trifecta" system to boost pressure
 - 325 psi
 - Limits standard operation pressure to ca. 260 psi
 - Higher pressures require auxiliary high pressure supply
- Flow control system to gasifier
 - Pressure regulator
 - Control valve
 - Coriolis flowmeter





Gasifier Specifications

Parameter	Typical	Max.
Pressure (bar)	18	31
Temperature (°C)	1425	1700
Slurry feed rate (lit/h)	50	150
Coal feed rate (kg/h dry)	30	80
Thermal input (kWth)	220	600
Slurry concentration (wt%)	59	65
Oxygen feed rate (kg/h)	35	150
Syngas production (m ³ /h dry)	50	150





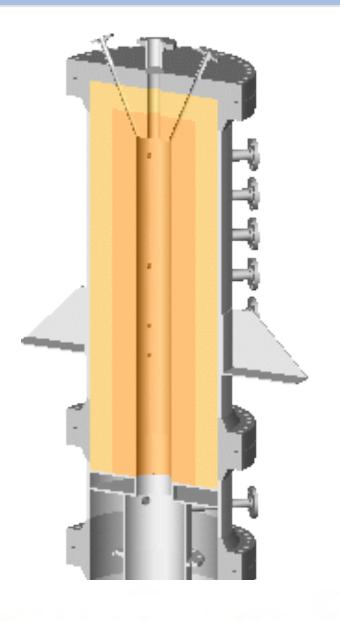
Reactor Details

Reactor dimensions

- 30 inch (0.75 m) pressure vessel
- 8.5 inch (0.22 m) reactor ID
- 60 inch (1.5 m) reactor length
- Designed to identify development of gas and condensed phases as coal undergoes conversion

Sample ports

- Twelve opposing 2 inch (5 cm) ports at six levels for sampling, optical diagnostics
- Two additional 2 inch (5 cm) ports at burner level
- Six 1 inch (2.5 cm) ports for temperature/ pressure measurement





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The Easy Stuff

Inputs

- Slurry flow rate and concentration
- Coal composition
- Oxygen feed rate
- Purge flow rates

Temperatures

- Five B-type thermocouples along length of reactor
- Additional thermocouples in quench, on shell, etc.

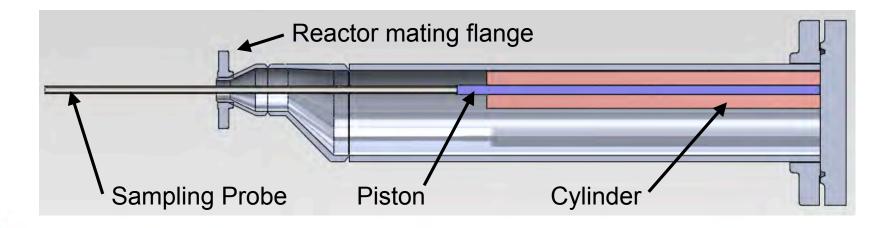
Syngas composition

- Analysis after gas has been quenched, cooled, depressurized
- Solids composition
 - Char caught in filters, slag caught in slag trap
 - Analyzed only after system is depressurized



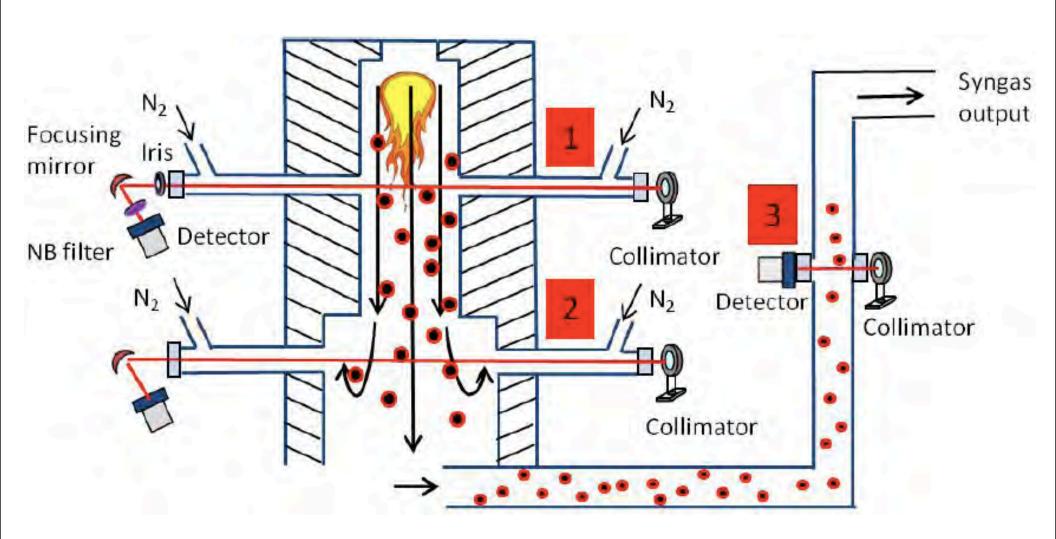
Extractive Sampling

- Cooled probe for gas sampling within reactor chamber
- Moveable piston will allow quick positioning from wall to centerline of reactor
- Safety systems integrated with gasifier control system
- Can be installed at any of five locations down length of reactor
- Modification of system will allow deposition of condensed-phase material onto probe



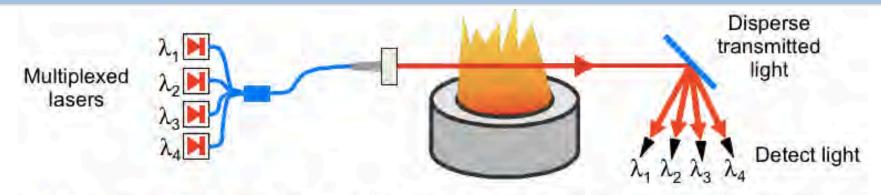


Measurement Locations for Stanford TDL Sensor Project

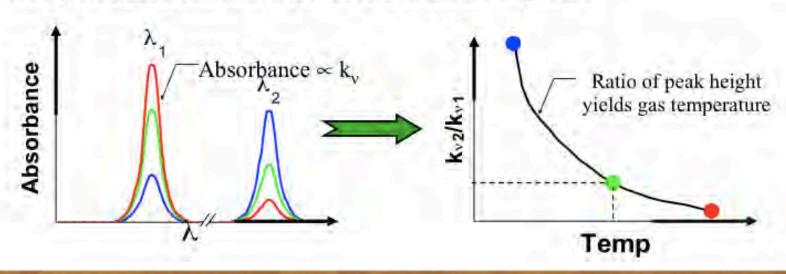




Absorption Fundamentals: Wavelength-Multiplexed Tunable Diode Laser Sensing

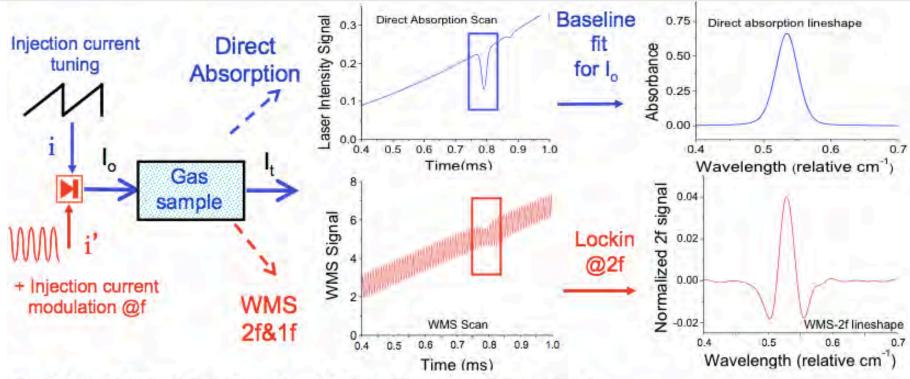


- Absorption of laser light by molecular transitions in the combustion gases
 - Beer's law: Transmission = $I/I_o = e^{-kL}$
 - Absorption coefficient k = f(temperature, pressure, gas composition)
- Ratio of absorbance on two molecular transitions yields gas temperature
- Multiplex additional lasers for more combustion species





Absorption Fundamentals: Scanned Direct Absorption and Wavelength Modulation Spectroscopy

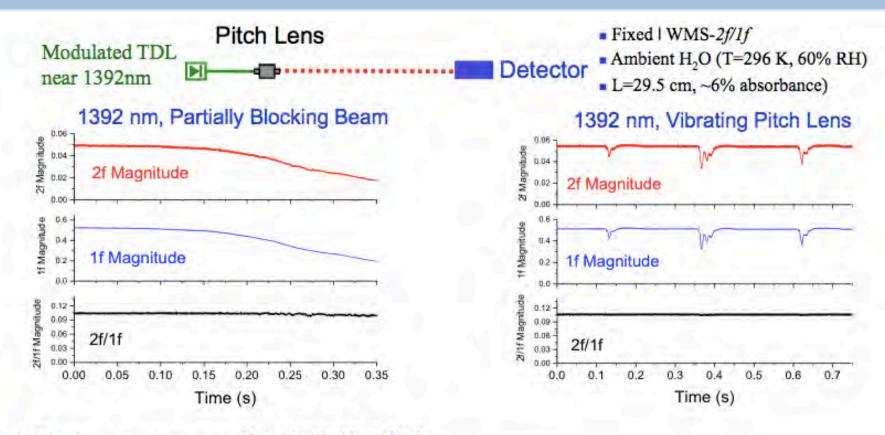


- Direct absorption: Simpler, if absorption is strong enough
- WMS: More sensitive especially for small signals (near zero baseline)
 - Ratio of two WMS-2f signals provides T (same as direct absorption)
 - WMS with TDLs has improved noise rejection (especially for non-absorption losses)
 - WMS also produces intensity modulation @1f
 - Since both 2f and 1f signals are proportional to I; 2f/1f independent of optical losses





Absorption Fundamentals: Demonstration that Normalization of WMS Improves Signal-to-Noise Ratio



- Demonstrate normalized WMS-2f/1f
 - No loss of signal when beam attenuated (e.g., scattering losses)
 - No loss of signal when optical alignment is spoiled by vibration
- Normalized WMS-2f/1f signals free from window fouling and particulate loading



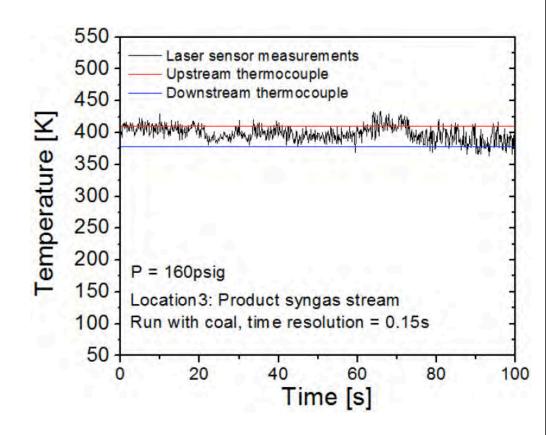


TDL Sensor Results at Position 3

Laser Transmission vs. Pressure

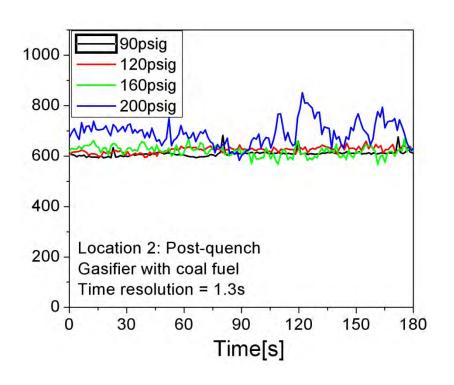
0.1 Location 3 Data 40 60 80 100 120 140 160 Pressure [psig]

Measured Temperature at 160 psi





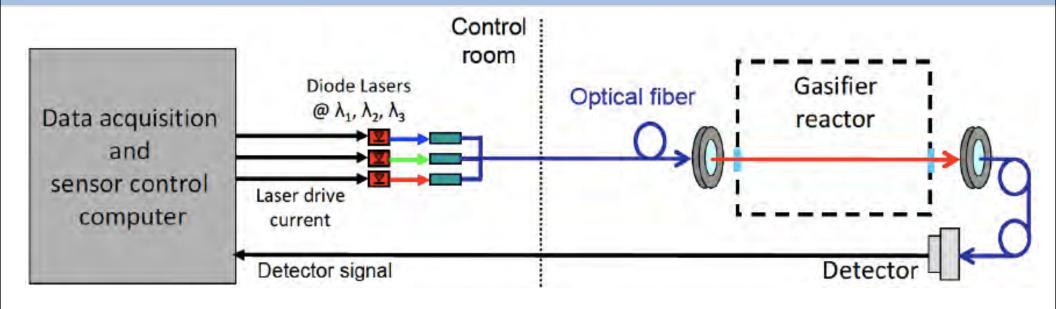
TDL Sensing at Position 2



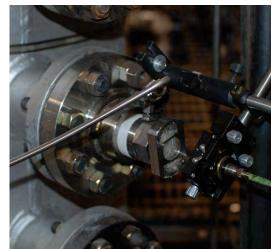


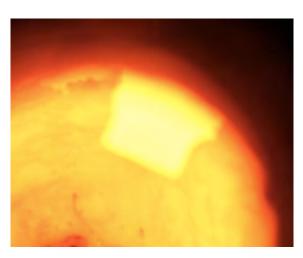
- High SNR, time-resolved measurements of T
- Normalized WMS accounts for varying transmission
- Measured T at reactor pressures of 90, 120 and 160psig stable
- Measured T at 200 psig identifies potential spray splashback instabilities

TDL-Based Measurements within Reactor "Core" (Position 1)









Validation Data Summary

Conditions

- To 500 kWth
 - -1.5 ton/day coal
- 440 psia (30 atm) pressure
 - Typically operate at psia
- Temp to 3100°F (1700°C)
 - Typically 2400-2600°F
- Various fuels
 - Pittsburgh #8
 - Illinois #6
 - Utah Sufco
 - Texas Lignite
 - Petcoke

Measurements

- Wall temperature
 - -5 positions
- Syngas composition
 - Post-quench
 - Pre-quench
- Reactor temperature
 - Integrated TDL-based
- Internal gas composition
 - Extractive sampling
 - Integrated TDL-based
- Internal condensed-phase
 - Extractive sampling



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Injector Cold Flow Test System

- Identification of injector performance
 - Uniformity
 - Spray angle
 - Droplet size
- Full scale model
 - Uses same injector as actual reactor
 - Air instead of oxygen
 - Water instead of slurry
- Pressurized system (to 5 bar) under development
- Analytical methods under development





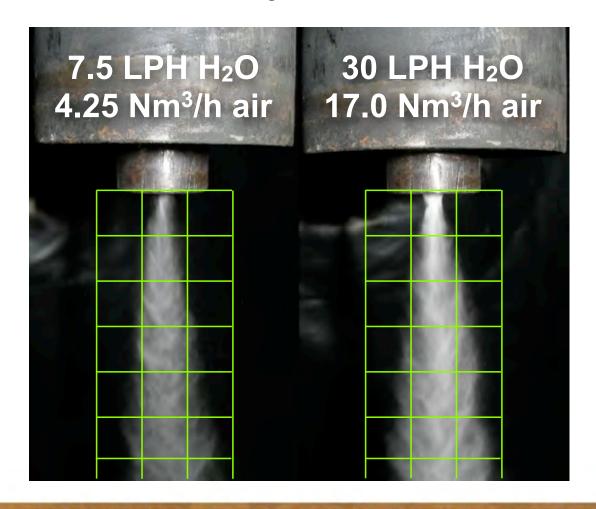


Flow Rate

Flow rates of air and water adjusted simultaneously to maintain air/water ratio

Air pressure drop 2.8 bar

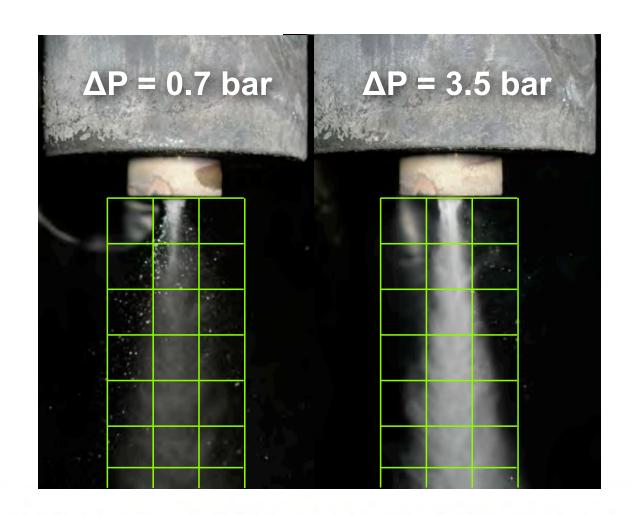
45 degree nozzle





Pressure Drop

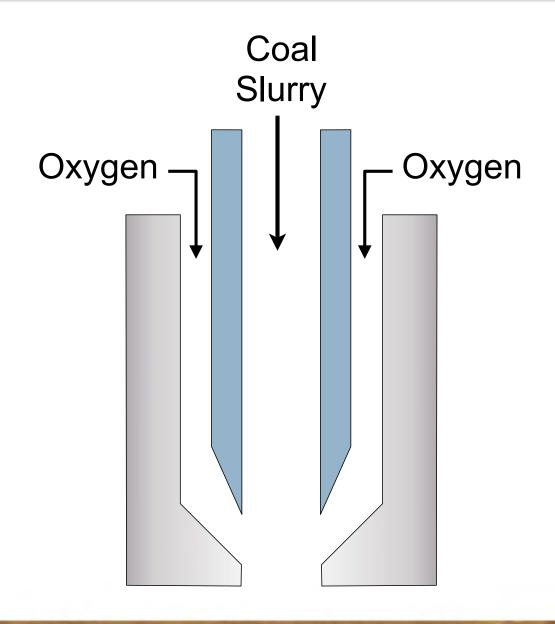
Both cases have 30 LPH water feed, 17 Nm³/h air feed 65 degree nozzle





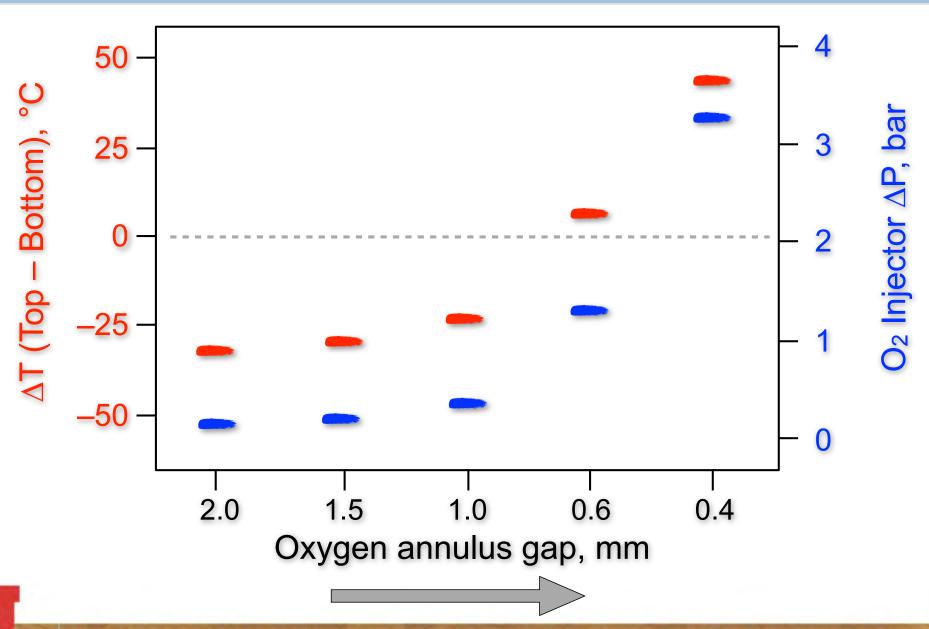
Adjustable Injector Tip







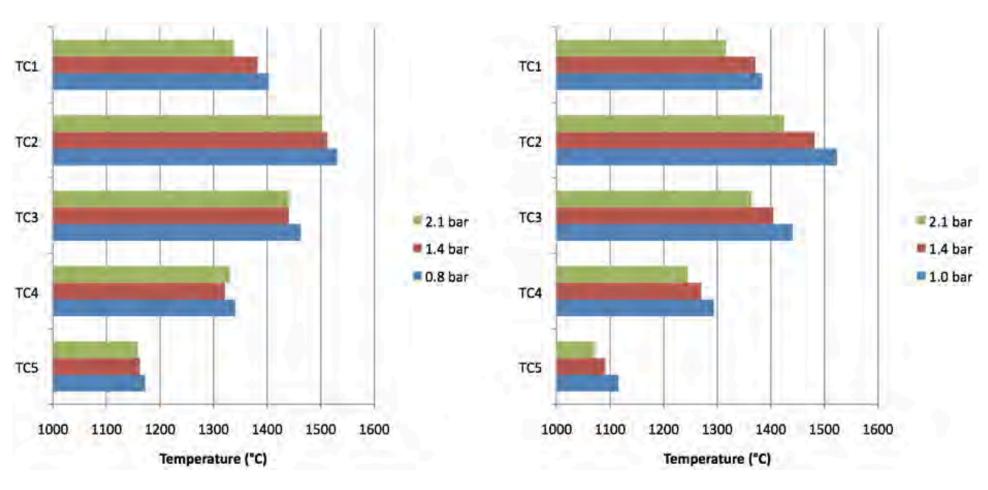
Performance vs. Injector Gap



Temperature Profile

45° nozzle angle (Day 1)





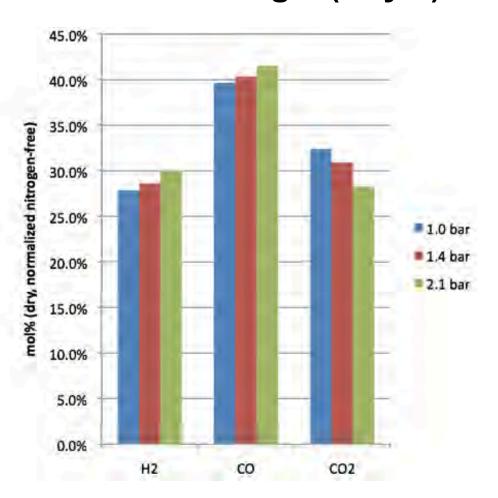


Syngas Composition

45° nozzle angle (Day 1)

45.0% 40.0% 35.0% mol% (dry, normalized nitrogen-free) 30.0% 25.0% 0.8 bar 20.0% 2.1 bar 15.0% 10.0% 5.0% 0.0% H2 co CO2

65° nozzle angle (Day 2)





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Uncertainty Considerations

- <u>Temperatures:</u> Thermocouple junction located within wall, approx. 0.5 inch from refractory face, to extend thermocouple life
- Gas composition: Cooling within extractive probe may affect gas composition due to:
 - Changes in gas equilibrium composition at lower temperatures
 - Absorption of minor constituents (sulfur compounds, ammonia) by condensed water
 - Condensation of e.g., polyaromatic hydrocarbons as gas is cooled



Uncertainty Considerations (2)

- <u>Condensed-Phase Material</u>: Difficult to obtain instantaneous compositions. Must be aggregate over time.
- All Data: Fluctuations on various time scales need to be quantified
 - 2 seconds
 - 20 minutes
 - Day-to-day



Conclusions

- Acquisition of data within reaction zone of pressurized gasifier very challenging
- Gasifier performance strongly tied to injector design and efficiency of fuel distribution
- Optical techniques offer unique opportunity for real-time non-invasive sampling
- Quantification of data variation and associated uncertainty is important if data is to be used for validation of simulations



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